# STRUCTURAL AND OPTICAL STUDIES OF GLYCINE BASED SINGLE CRYSTALS - A NONLINEAR OPTICAL MATERIAL

V.Vidhya<sup>1</sup>,R. Muraleedharan<sup>1</sup>, Ramajothi<sup>2</sup> and G.Vinitha<sup>3</sup>

<sup>1\*</sup> Assistant Professor, Department of Physics, Ponnaiyah Ramajayam Institute of Science and Technology [PRIST], Tamil Nadu, India <sup>2</sup> Department of Physics, Anna University, Chennai India <sup>3</sup>School of Advanced Sciences, VIT Chennai, India \*Corresponding author:jrramajothi@gmail.com

## Abstract

The glycine based single crystals such as glycine zinc sulphate (GZS) and di(glycine) barium chloride (DGBC) was synthesized and single crystals are grown by solution growth technique. The unit cell parameters of grown crystals have been analyzed by single crystal XRD. The optical properties were studied using UV-Vis and photoluminescence spectroscopies. The TG/DTA analyzes were performed and the melting point of the sample was measured. The study of microhardness was measured on a prominent plane to determine mechanical strength using Vicker's hardness test. The NLO properties measurements were carried out using Q-switched Nd: YAG laser.

Keyword: Single crystals, unit cell, microhardness, band gap, photoluminescence.

# 1. Introduction

The NLO material plays an important role in photonics, laser technologyand optical modulation since it can produce a high value of the original frequency[1-3]. To realize many of these NLO applications, materials with large second order nonlinearities, short transparency cutoff wavelength and stable physicochemical properties are needed[4-7]. Due to their possible applications in the optoelectronic technologies, nonlinear optical materials of third order property have received much attention [8-10]. Thirdorder nonlinear optical materials with low nonlinear absorption but strong nonlinear refraction have attracted considerable attention because of their potential use in the optical signal processing device[11-13]. The structural, thermal, and optical properties of a GZS and DGBC semiorganic nonlinear single optical crystal were investigated.

# 2. Experimental Techniques

# 2.1 Synthesis and crystal growth

The GZS was synthesized from glycine and zinc sulfate in the ratio 1:1. The reaction as follows

 $ZnSO_4 \cdot 7H_2O + CH_2NH_2COOH \rightarrow Zn[CH_2 NH_2COOH] SO_4 \cdot 7H_2O$ 

The DGBC salt was synthesized with glycine and barium chloride in the ratio 2:1. The reaction is taken as  $2(NH_2-CH_2-COOH) + BaCl_2 + 2H_2O \rightarrow Ba (NH_2-CH_2-COOH)_2Cl_2 + 2H_2O$ 

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The molecular structure of GZS and DGBC was shown in Fig. 1(a) and (b) respectively.



Glycine zinc sulfate

#### Fig. 1 (a) Molecular structure of GZS





#### 2.2 Solubility Studies

The solubility test for GZS and DGBC was carried with fine different temperatures (35, 40, 45, 50, 55°C) from the solubility curve highly soluble in distilled water is observed (Fig. 2).



Fig. 2 Solubility curve of GZS and DGBC

# 2.3 Growth of single crystals

The synthesized GZS salt was dissolved in distilled water then allowed for crystallization and a good quality single crystal with dimensions of  $14 \times 10 \times 5$  mm<sup>3</sup> obtaine within a month (Fig. 3). Similarly, the DGBC solutions were prepared then kept for crystallization. The good quality single crystals with a size of  $30 \times 15 \times 10$  mm<sup>3</sup> are grown in 25 days (Fig. 4)



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Fig. 3. As grown single crystals of GZS



Fig. 4. As grown single crystal of DGBC

# **3.Result and Dissucion**

# 3.1 Single Crystal XRD Analysis

The grown crystals (GZS and DGBC) were subjected to single crystal XRD analysis using ENRAF NONIUS CAD-4 X-ray diffractometer with MoK $\alpha$  ( $\lambda = 0.717$  Å) radiation. The GZS and DGBC belong toTriclinic P and Orthorhombic P crystal system respectively (Table 1).

Crystal	GZS	DGBC	
a(Å)	5.98	8.27	
b(Å)	6.84	9.32	
c(Å)	13.35	14.90	
α	85.12	90	
β	83.21	90	
γ	82.84	90	
$V(Å)^3$	537	1149	

## Table 1. Single crystal XRD of GZS and DGBC

#### **3.2. UV-Vis Spectral Study**

The UV-Vis spectral transmission analysis of GZS and DGBC was carried out using Perkin Elmer Lambda UV-Vis spectrophotometer (Fig. 5). The lower cutoff wavelength for GZS and DGBC is 190 nm and has a good optical transmission window (190-1100 nm) throughout the UV, Visible, and Near IR region that allows for optoelectronic applications.



Fig. 5 UV-Vis transmittance spectrum of GZS and DGBC

The band gap of GZS and DGBC crystals was determined by plotting  $(\alpha hv)^2 vs$  (hv) and then band gap energy is found to be 6.4 and 6.8 eV (Fig.6 and Fig.7). The extinction coefficient of glycine based single crystals was calculated from the relation.

$$k = \frac{\alpha \lambda}{4\pi}$$

The extinction coefficient of GZS and DGBC was decreasing with the increasing photon energy (Fig.

8).

#### Fig.6Tauc's plot of GZS crystal





Fig. 8Photon energy Vs extinction coefficient

#### 3.3 Photoluminescence Spectrum

The photoluminescence study of GZS (Fig. 9) and DGBC (Fig. 10) was carried out using a Cary eclipse photoluminescence spectroscopy with excitation wavelength of 220 nm. In both crystals the maximum violet emission was observed at the peak with energy of 3eV and the peak at 465 nm shows the weak blue emission (2.7eV). The peaks at 270 nm, 325 nm and 360 nm emit the weak violet emission corresponding to the energy of 4.6 eV, 3.8 eV and 3.4 eV respectively.



Fig.9 Photoluminescence (PL) spectrum of GZS



Fig. 10 Photoluminescence (PL) spectrum of DGBC

#### 3. 4. FT-IR spectral analysis

The FT-IR spectrum of GZS and DGBC was recorded in the range of 400-4000 cm<sup>-1</sup> by KBr pellet technique (Fig. 11). The bands at 3788 cm<sup>-1</sup> and 1474 cm<sup>-1</sup> due to NH asymmetric stretching and COO<sup>-</sup> stretching [7]. The band at 3003 cm<sup>-1</sup>established the presence of CH group [8]. The bands at 1640 cm<sup>-1</sup> and 1420 cm<sup>-1</sup>established the presence of NH<sub>3</sub><sup>+</sup> and C=O symmetric stretching. The band 1108 cm<sup>-1</sup> confirm the presence of C-C stretching. The band at 619 cm<sup>-1</sup> is assigned to the COO- out of plane bending vibration [9]. The vibration of CH<sub>2</sub> group appears at 1474 cm<sup>-1</sup>. The characteristic stretching vibrations of SO<sub>4</sub><sup>-</sup> group appear at 892 and 664 cm<sup>-1</sup>. The carboxylate group (COO<sup>-</sup>) asymmetric and symmetric stretching vibrations appear at 1640 and 1420cm<sup>-1</sup>. The peak observed at 1312 cm<sup>-1</sup> is due to the CH<sub>2</sub> wagging vibration. The frequencies at 892 cm<sup>-1</sup> are assigned to the carboxylate group. The presence of carboxylate ion and ammonium ion clearly indicated that glycine molecule exists in the zwitter ionic form in GZS [12]. In DGBC a peak at 2583 indicates the presence of NH<sub>3</sub><sup>+</sup> stretching vibrations. A peak at 1572 cm<sup>-1</sup>confirm NH<sub>3</sub><sup>+</sup> asymmetric bending vibration in a primary amine group. The band 1334 cm<sup>-1</sup>established the presence of C-N stretching vibration. The peaks at 1115 cm<sup>-1</sup> and 896 cm<sup>-1</sup> are due to the NH<sub>3</sub><sup>+</sup> group of glycine moleculeand CCN stretching groups respectively.



Fig. 11FT-IR spectrum of GZS and DGBC

#### 3.5.TG/DTA analyzes

The TG/DTA of GZS and DGBC was performed using SII Nanotechnology TG/DTA 6200. The TGA curve (Fig.12) of GZS shows good thermal stability up to 122°C as there is 14% of weight loss is due to the removal of occluded and absorbed water molecules occurring up to 85°C. The DTA curve shows that the melting point of GZS is 123°C. The TGA curve (Fig. 13) of DGBC has good thermal stability up to 190°C as there is no weight loss below that temperature and weight loss of about 30% in the temperature range 200-320°C. The DTA curve shows the peak at 190°C to be the DGBC melting point.



Fig. 13 TG-DTA curves of DGBC

#### 3. 6 Measurement of Microhardness

Minimal Andrews studies for GZS and DGBC were condition and the well grown plane using a SHIMADZU HMV-200 fitted with a Vicker's pyramidal indenter(Fig. 14). The GZS and DGBC work hardening coefficient (Fig. 15) is 1.94 and 1.97 which indicates that both crystals belong to the category of soft material.



Fig. 14 Variation of hardness with load of GZS and DGBC



Fig. 15 Variation of log d Vs log P of GZS and DGBC.

# 3.7Determinatio of Second Harmonic Generation Efficiency

The second order nonlinear optical property (SHG) of GZS crystal was determined by Kurtz and Perry technique[13-15]. A Q-switched Nd : YAG laser beam was allowed to hit the sample, emitting a fundamental wavelength of 1064 nm (pulse width 8 ns). The SHG in the crystalline sample was confirmed by green radiation emission (532 nm) from the sample. The SHG efficiency of GZS was found to be 0.7 times of KDP. The GZS belongs to non-cento symmetric space group which exhibits SHG efficiencywhereas DGBC belongs to center symmetric space group which does not exhibit second order NLO phenomena.

# 3.8 THG Measurement Using Z-Scan Technique

The Z-scan experiment was carried out for GZS and DGBC using 50 mW Diode pumped Nd: YAG laser with a wavelength of 532 nm [16, 17]. The closed and open aperture of the third order nonlinear optical

susceptibility  $\chi^{(3)}$  for GZS and DGBC are shown in Fig. 16 and 17 respectively. The ratio of an open and closed aperture of third order nonlinear optical susceptibility for GZS and DGBC was shown in Fig.18. The third order nonlinear parameters of GZS and DGBC are shown in Table.2.







Fig. 18 Ratio of open and closed aperture of GZS and DGBC

Parameters	GZS	DGBC
50 mW Diode pumped Nd:YAG Laser ( $\lambda$ )	532 nm	532 nm
Lens focal length (f)	3.5 cm	3.5 cm
Optical path length	1 mm	1 mm
Beam radius of the aperture (wa)	15 mm	15 mm
Aperture radius (ra) for closed aperture	2 mm	2 mm
Linear refractive index	1.68	1.629
Nonlinear refractive index (n <sub>2</sub> )	5.017×10 <sup>-</sup> <sup>8</sup> cm <sup>2</sup> /W	4.346×10 <sup>-</sup> <sup>8</sup> cm <sup>2</sup> /W
Nonlinear absorption coefficient ( $\beta$ )	0.018×10 <sup>-4</sup> cm/W	0.023×10 <sup>-4</sup> cm/W
Real part of the third order susceptibility[Re( $\chi^3$ )]	10.030×10 <sup>-6</sup> esu	9.818 ×10 <sup>-6</sup> esu
Imaginary part of the third ordersusceptibility $[Im(\chi^3)]$	0.110×10 <sup>-6</sup> esu	0.145×10 <sup>-6</sup> esu
Third order nonlinear optical susceptibility $(\chi^3)$	10.030 ×10 <sup>-6</sup> esu	9.819×10 <sup>-6</sup> esu

Table. 2. Nonlinear properties of GZS and DGBC single crystals

# 4. Conclusion

The GZS and DGBC single crystals were grown by evaporation method and single crystal XRD analysis confirmed that GZS and DGBC crystals belong respectively to the triclinic P and orthorhombic P crystal systems. FTIR analysis confirmed the molecular structure of the synthesized material. The transmission spectrum reveals that GZS and DGBC have a lower cutoff wavelength of 190 nm. The TG/DTA analysis shows that the grown crystal have good thermal stability and both crystals are belongs to soft material category. The SHG of GZS crystal was found to be 0.7 times that of KDP. The third-order nonlinear susceptibility  $\chi^{(3)}$  for GZS and DGBC crystals are found to be 10.030 and 9.819×10<sup>-6</sup>esu respectively.

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